Structure-activity Relationships of Pyripyropenes Fungal Acyl-CoA: Cholesterol Acyltransferase Inhibitors

Sir:

Acyl-CoA: cholesterol acyltransferase (ACAT) has been believed to be an intriguing target of inhibition as a new type of atheroscrelotic agents. Pyripyropenes A to D isolated from the fermentation broth of Aspergillus fumigatus FO-1289 represent the most potent inhibitory activity in naturally occurring ACAT inhibitors. Pyripyropenes inhibit ACAT activity in nanomoler level of IC₅₀ value.^{1~3)} Pyripyropene A (1) proved to be orally active for reducing cholesterol absorption in a hamster model.¹⁾ Recently, the stereochemistry of 1 was determined as shown in Fig. 1.4) Among the four compounds, pyripyropene C (2) showed the most potent inhibition, followed by 1. The structural difference between them lies in the 7-O-acyl group; 1 has an acetyl residue, and 2 does a longer propionyl one, suggesting that synthetic replacement at the 7-O-acyl residue is to be investigated. Here we describe the synthesis and structure-activity relationships of 7-O-acyl group.

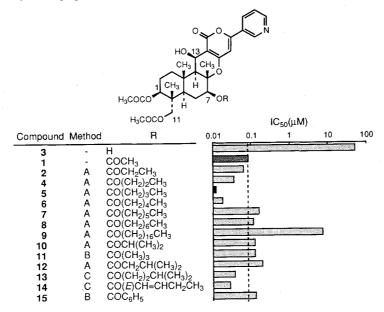
Hydrolysis of pyripyropene A with 1,8-diazabicyclo-[5,4,0]undec-7-ene (DBU) in 80% methanol gave 7-OH derivative (3) in 52% yield. 3 was acylated by acyl anhydride and triethylamine with 4-dimethylaminopyridine (DMAP) in dichloromethane (Method A), or with acyl chloride and triethylamine in dichloromethane (Method B). Some derivatives were also prepared by appropriate carboxylic acid and DCC-DMAP in dichloromethane (Method C). The structures of the derivatives were determined by NMR analysis and MS. ACAT inhibitory activity was assessed using rat liver microsomes as an enzyme source according to our established

method.5,6)

When the ACAT inhibitory activity was compared in derivatives having a linear acyl group $(1, 2 \text{ and } 4 \sim 9)$, one with longer carbon length up to 5 increased the inhibitory activity, but another with longer than 7 carbons length decreased the activity. As a result, the *n*-valeryl derivative with 5 carbons length (5) showed the best ACAT inhibition. [Analytical data of 5: HRFAB-MS, found $626.2945 (M+1)^+$ calcd 626.2964, ¹H NMR $(270 \text{ MHz}, \text{CDCl}_3)$, $\delta 8.99 (1\text{H}, \text{d}, J = 1.7 \text{Hz}, 2'' - \text{H})$, 8.68(1H, dd, J=1.7, 5.0 Hz, 6''-H), 8.09 (1H, dt, J=2, 8 Hz,4"-H), 7.40 (1H, dd, J = 5.0, 8 Hz, 5"-H), 6.39 (1H, s, 5'-H), 5.01 (1H, m, 7-H), 4.99 (1H, d, J = 3.0 Hz, 13-H), 4.78 (1H, dd, J = 5.3, 10.9 Hz, 1-H), 3.80 (1H, d, J =11.9 Hz, 11-H), 3.68 (1H, d, J = 11.9 Hz, 11'-H), 2.90 (1H, d, J = 11.9 Hz)br s, 13-OH), 2.40 (2H, dt, J=2.0, 7.3 Hz, $COCH_2CH_2$ -CH₂CH₃), 2.16 (1H, m, 3-H), 2.08 (3H, s, 11-COCH₃), 2.04 (3H, s, 1-COCH₃), $1.9 \sim 1.7$ (2H, m, 2 and 2'-H), 1.68 (3H, s, 14-CH₃), $1.7 \sim 1.5$ (6H, m, 5, 8, 8', 9 and COCH₂CH₂CH₂CH₃), 1.43 (3H, s, 12-CH₃), 1.38 (3H, m, 3'-H and COCH₂CH₂CH₂CH₃), 0.96 (3H, t, $J = 7.4 \,\mathrm{Hz}$, COCH₂CH₂CH₂CH₃), 0.88 (3H, s, 15-CH₃).] Introduction of branched acyl groups ($10 \sim 13$) resulted in decrease of the activity in comparison with the corresponding linear acyl groups $(4 \sim 6)$, even though the isovaleryl derivative with 5 carbons (13) still showed stronger activity than 1. Similarly, the alkene 14 with a 5 carbons chain also showed stronger activity than 1. The activity of the aromatic derivative (15) was almost the same as that of the linear acyl derivatives (7 and 8) and the branched acyl derivatives $(10 \sim 12)$.

In summary, the 7-O-acyl moiety was modified with several acyl groups in order to find more potent ACAT inhibitors than 1. Among 15 derivatives, 5 having an *n*-valeryl group showed the most potent ACAT inhibi-

Fig. 1. Structures, synthetic preparation methods and in vitro ACAT inhibitory activity of pyripyropene derivatives.



tion with an IC_{50} value of 13 nm, indicating 7-fold stronger activity than 1. This result suggests that 7-O-acyl moiety of pyripyropene plays an important role to show the ACAT inhibitory activity.

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